



Karl Fischer Analysis of Fuel Ethanol

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Today's Session

- Review KF reaction
- Volumetric vs. coulometric
- Choosing the right sample size
- Common hurdles
- Q&A
- Additional Resources

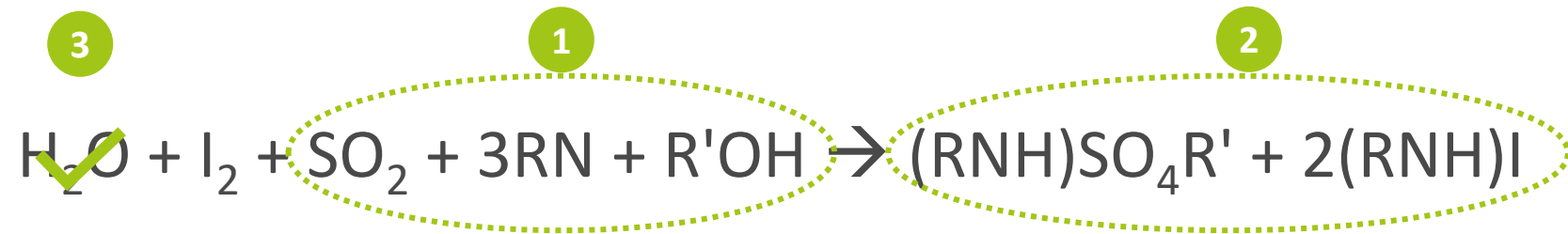


Karl Fischer Reaction

- 1 Alkylsulfite reactive intermediate

- 2 Sulfite group oxidized to sulfate group by iodine

- 3 One mole of water is consumed

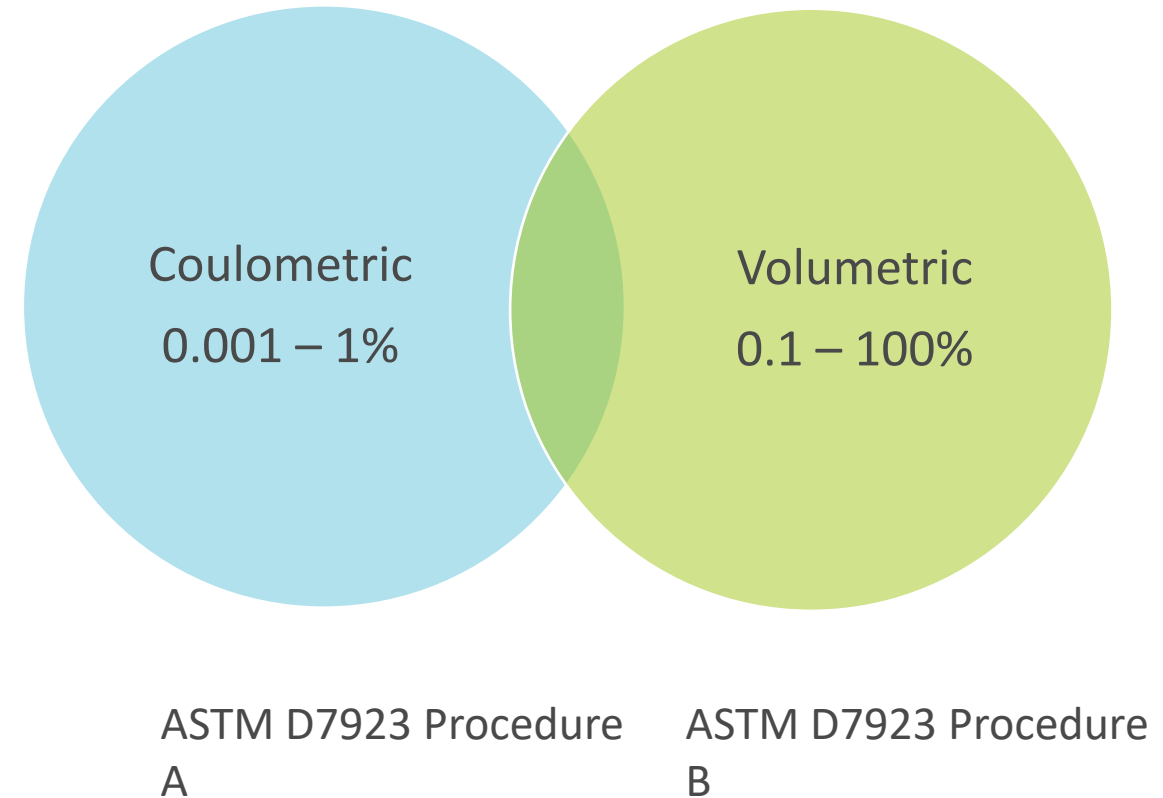


By knowing volume of iodine consumed, water concentration is directly determined

Karl Fischer Techniques

Choose a technique based on:

- Expected moisture concentration
- Available sample
- Sample solubility



Coulometric Karl Fischer

- 0.001% - 1% range
- Iodine generated from iodide containing reagent using current
- Generating current switched off with detection of slight excess of free iodine
- Amount of iodine generated is known from charge and time

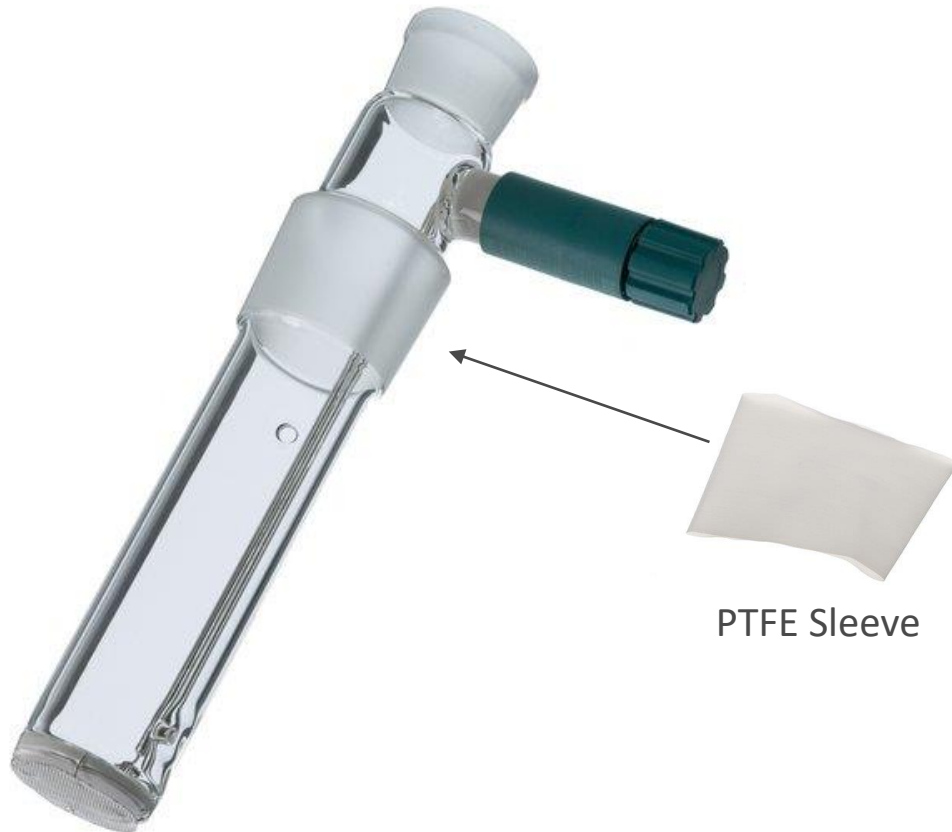


Double Platinum Electrode



- Indicates excess iodine in volumetric and coulometric systems
- Coulometric systems – use PTFE sleeve; not grease
- Pins should be parallel
- Clean as needed and store dry (or in vessel)
- Damaged electrode or cable can appear as overtitration

Generator Electrode



- Generates iodine from iodide solutions by applying current
- Handle delicately, do not bend mesh or connections
- Dirty electrode:
 - Sluggish response
 - Decreased iodine generation
 - Mesh/diaphragm turns gray
- Cleaning procedure:
 - Rinse with water or solvent
 - 50% nitric acid soak (10-15 minutes)
 - Rinse with water, then methanol, then dry

Volumetric Karl Fischer

- 0.1 – 100% range
- Solvent solution consists of methanol
- Titrate with iodine reagent
- Fast titrations
- Solids, liquids, gases
- Very modifiable system
 - Solubility promoters
 - Homogenizer
 - Temperature



Cleaning Karl Fischer Titration Vessels

Coulometric vessels are
single piece with glass
joints

Volumetric vessels have
separate parts and often
use o-rings

- 1 Empty vessel and rinse with solvent

- 2 Wash with soap and water

- 3 Rinse with solvent

- 4 Soak in 50% nitric acid for 10-15 minutes

- 5 Rinse with water

- 6 Rinse with solvent

- 7 Dry in oven < 60°C



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Procedure A - Coulometric

- Water up to 2% by mass
- Pre-titrate or condition KF vessel
- Rinse syringe with sample
- Withdraw appropriate sample amount
- Invert the syringe to eject air
- Wipe excess needle from syringe
- Obtain mass of syringe containing sample to ± 0.1 mg
- Inject sample below the level of the KF solution
- Withdraw and weigh syringe to ± 0.1 mg

TABLE 1 Recommended Sample Size (Coulometric)

Expected Water Content (mass percent)	Sample Size (g)
0 to 0.2	3 to 5
0.2 to 0.5	1 to 2
0.5 to 1.0	0.5 to 1.0
1.0 to 2.0	0.5
>2.0	Please use Procedure B



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Procedure B - Volumetric

- Water up to 5.4% by mass
- Prepare buret and tubing
- Pre-titrate or condition KF vessel
- Rinse glass gas-tight syringe with sample
- Withdraw appropriate sample amount
- Invert the syringe to eject air
- Wipe excess needle from syringe
- Obtain mass of syringe containing sample to ± 0.1 mg
- Inject sample below the level of the KF solution
- Withdraw and weigh syringe to ± 0.1 mg

TABLE 3 Recommended Sample Size (Coulometric)

Expected Water Content (mass percent)	Sample size at titrant strength, H ₂ O at 2 mg/mL, (g)
0.2	2 to 5
0.5	0.8 to 4
1.0	0.4 to 2
1.5	0.25 to 1.5
>3.0	0.2 to 0.4

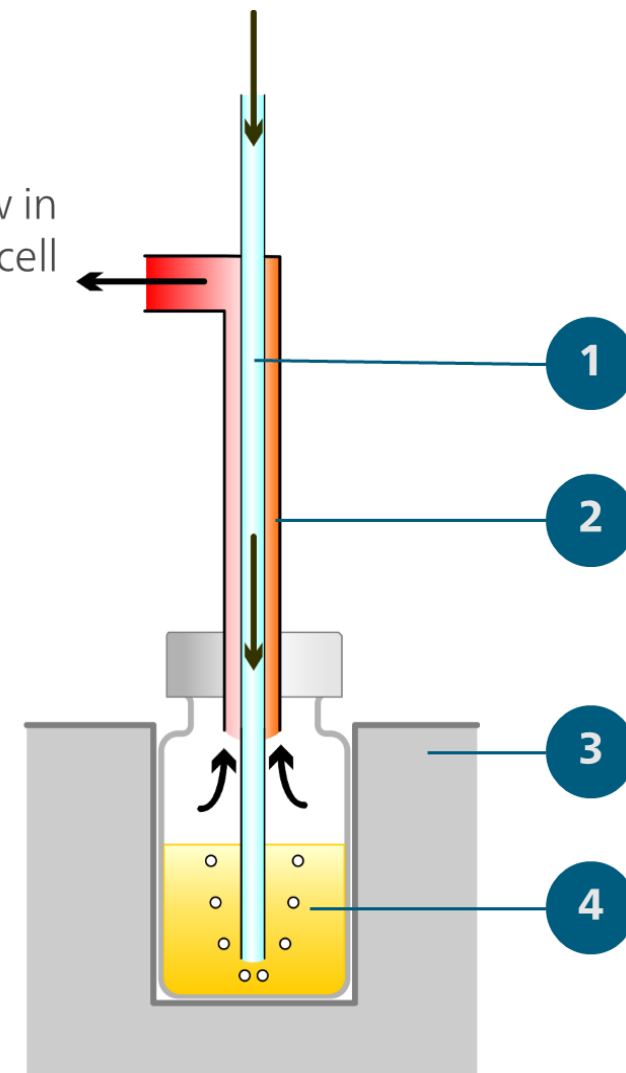


Karl Fischer Oven

- Corn oil
- DDG
- Isolates sample in sealed vial
- Low maintenance
- Can be automated
- Saves cost on reagents

- 1 Inlet needle
- 2 Outlet needle
- 3 Oven
- 4 Vial with sample in oven

Gas flow in titration cell



Oven Method Step by Step



- Conditioning before each determination
- Blank value of empty sample vial
- Absolute amount of water in the range of hundreds of μg
- Nitrogen or other dry and inert gas as carrier gas
- Temperature gradient to define optimal oven temperature
- Oven method can be fully automated

Quality Control

Water Standards:

- 0.1% or 1% recommended for coulometric
- 1% recommended for volumetric



Questions?