

# Process Water Analysis

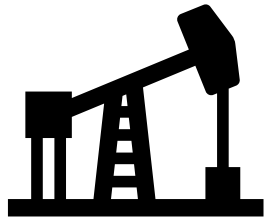
FELC 2024

Larry Tucker

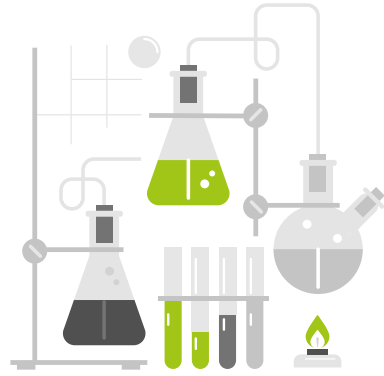
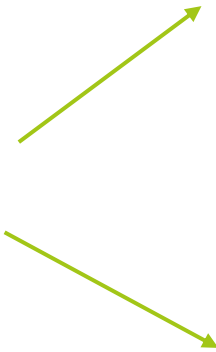
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# Water Usage



**WATER**



**UTILITIES**



**SLURRY TANKS  
AND LIQUIFICATION**



**FERMENTATION**

# Utility Water

At a minimum, boiler feedwater must be softened water for low pressure boilers and demineralized water for high pressure boilers. It must be free of oxygen and essentially free of hardness constituents and suspended solids.

Feedwater chemistry parameters to be controlled include dissolved solids, pH, dissolved oxygen, hardness, suspended solids, total organic carbon (TOC), oil, chlorides, sulfides, alkalinity, and acid- or base-forming tendencies.



Boiler  
Feedwater  
Guidance

Parameter	Concentration	Instrument
Cation Conductivity, $\mu\text{S}/\text{cm}$	<0.2	Conductivity
pH	9.0 - 9.6	pH
Dissolved Oxygen( $\text{O}_2$ ), $\mu\text{g}/\text{l}$	< 20	Oxygen Meter
Sodium (Na), $\mu\text{g}/\text{l}$	$\leq 2$	IC or ICP
Chloride (Cl), $\mu\text{g}/\text{l}$	$\leq 3$	IC
Silica ( $\text{SiO}_2$ ), $\mu\text{g}/\text{l}$	< 10	Colorimetric
Iron (Fe) total, $\mu\text{g}/\text{l}$	< 2	Color/ICP
Copper (Cu) total, $\mu\text{g}/\text{l}$	<2	Color/IC/ICP
TOC, $\mu\text{g}/\text{l}$	<100	TOC
Alkalinity	100 - 400 P alk	Titration

**Table 1**  
**Recommended Feedwater Limits**

Pressure, psig (MPa)	Drum Boilers						Once-Through Boilers		
	15 to 300 (0.10 to 2.07)	301 to 600 (2.08 to 4.14)	601 to 900 (4.14 to 6.21)	901 to 1000 (6.21 to 6.90)	1001 to 1500 (6.90 to 10.34)	>1500 (>10.34)	with AVT <sup>a</sup> All	AVT All	Oxygen Treatment All
pH, all-ferrous heaters	9.3 to 10.0	9.3 to 10.0	9.3 to 10.0	9.3 to 9.6	9.3 to 9.6	9.3 to 9.6	9.3 to 9.6	9.3 to 9.6	8.0 to 8.5
pH, copper- bearing heaters	8.8 to 9.2	8.8 to 9.2	8.8 to 9.2	8.8 to 9.2	8.8 to 9.2	8.8 to 9.2	9.0 to 9.3 <sup>b</sup>	8.8 to 9.2	N/A
Total hardness, as ppm CaCO <sub>3</sub> maximum	0.3	0.2	0.1	0.05	0.003	0.003	0.003	0.003	0.001
Oxygen, ppm maximum <sup>c</sup>	0.007	0.007	0.007	0.007	0.007	0.007	0.007 <sup>d</sup>	0.007	0.030 to 0.150
Iron, ppm maximum	0.1	0.04	0.02	0.02	0.01	0.01	0.01	0.010	0.005
Copper, ppm maximum	0.05	0.02	0.01	0.01	0.005	0.002	0.005	0.002	0.001
Organic, ppm TOC max.	1.0	1.0	0.5	0.2	0.2	0.2	0.2	0.200	0.200
Cation conductivity, μS/cm max.	—	—	—	—	0.5	0.2	0.2	0.15	0.15

<sup>a</sup> All-volatile treatment.

<sup>b</sup> AVT not recommended for copper-bearing cycles and associated low feedwater pH where the drum pressure is less than 400 psig.

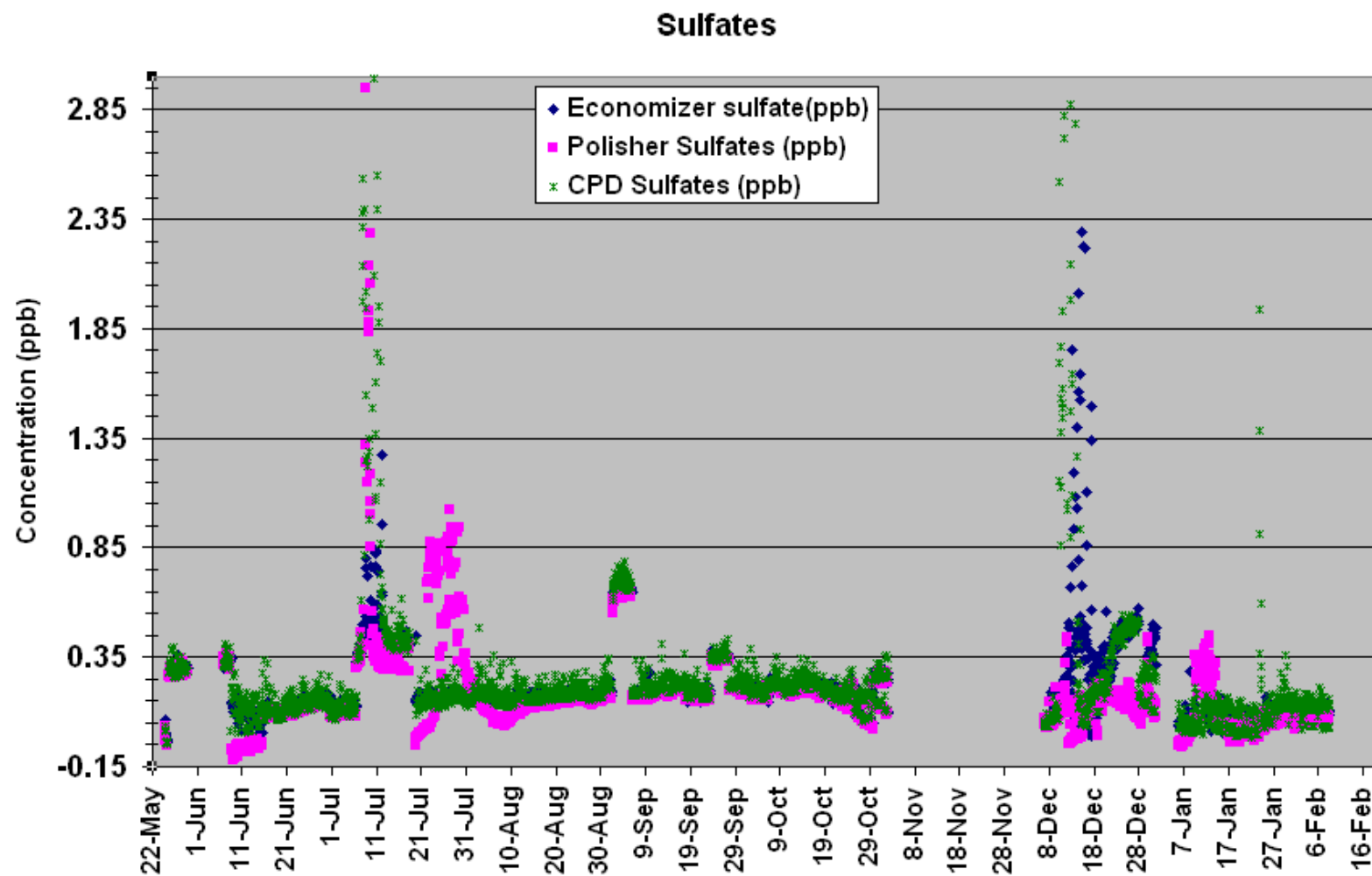
<sup>c</sup> By mechanical deaeration before chemical scavenging of residual.

<sup>d</sup> Dissolved oxygen for AVT(R); no oxygen scavenger used with AVT(O).

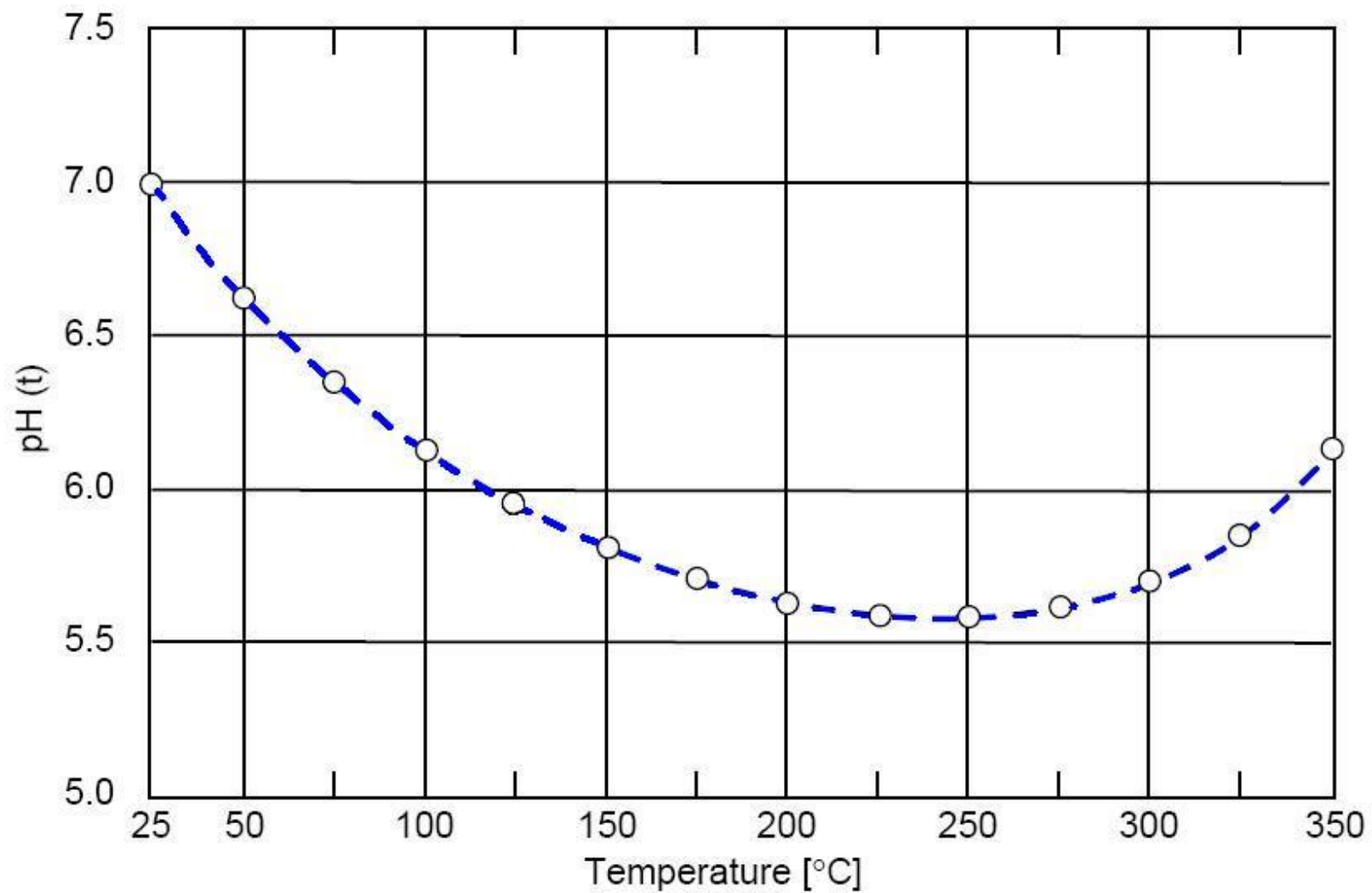
Note:

ppm = mg/kg

# Trending for Predictive Maintenance



# pH of Water At Temperature



# pH Control

Ammonia

Alkalizing Amines

**MONOETHANOLAMINE (MEA)**

**DIETHANOLAMINE (DEA)**

**TRIETHANOLAMINE (TEA)**

ETC.

Wastewater plant

**10 MG/L MEA, PH ~ 9.7 @ 25 °C**

**SHOCK LOAD TO BIO TREATMENT**





# Summary

Monitoring Plan

Efficient Operation

Control Corrosion

Run vs Repair

Economic Benefit



# Resources

## Analytical Methods for Water

ASTM COMMITTEE D19 WATER

STANDARD METHODS FOR THE EXAMINATION OF WATER AND WASTEWATER 24<sup>TH</sup> EDITION

THE B&W LEARNING CENTER [HTTPS://WWW.BABCOCK.COM/HOME/ABOUT/RESOURCES/LEARNING-CENTER/THE-IMPORTANCE-OF-BOILER-WATER-AND-STEAM-CHEMISTRY#:~:TEXT=AT%20A%20MINIMUM%2C%20BOILER%20FEEDWATER,HARDNESS%20CONSTITUENTS%20AND%20SUSPENDED%20SOLIDS](https://www.babcock.com/home/about/resources/learning-center/the-importance-of-boiler-water-and-steam-chemistry#:~:text=at%20a%20minimum%2c%20boiler%20feedwater,hardness%20constituents%20and%20suspended%20solids)

# Liquefaction & Fermentation

Regulating pH

Measuring pH



# pH value: Temperature dependence

Nernst equation:

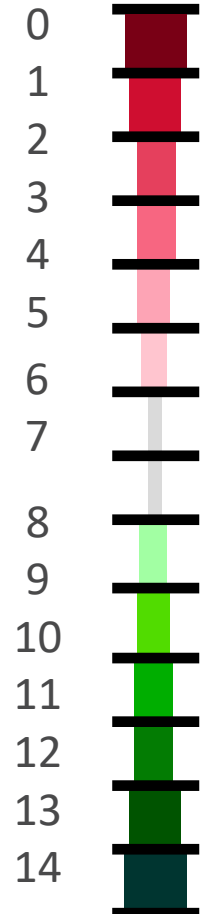
$$U = U_0 + \frac{2.303 \times R \times T}{z \times F} \times \log a_{H^+}$$

Correct measurement  
requires pH and  
temperature  
e.g. pH 5.23 (24.5 °C)

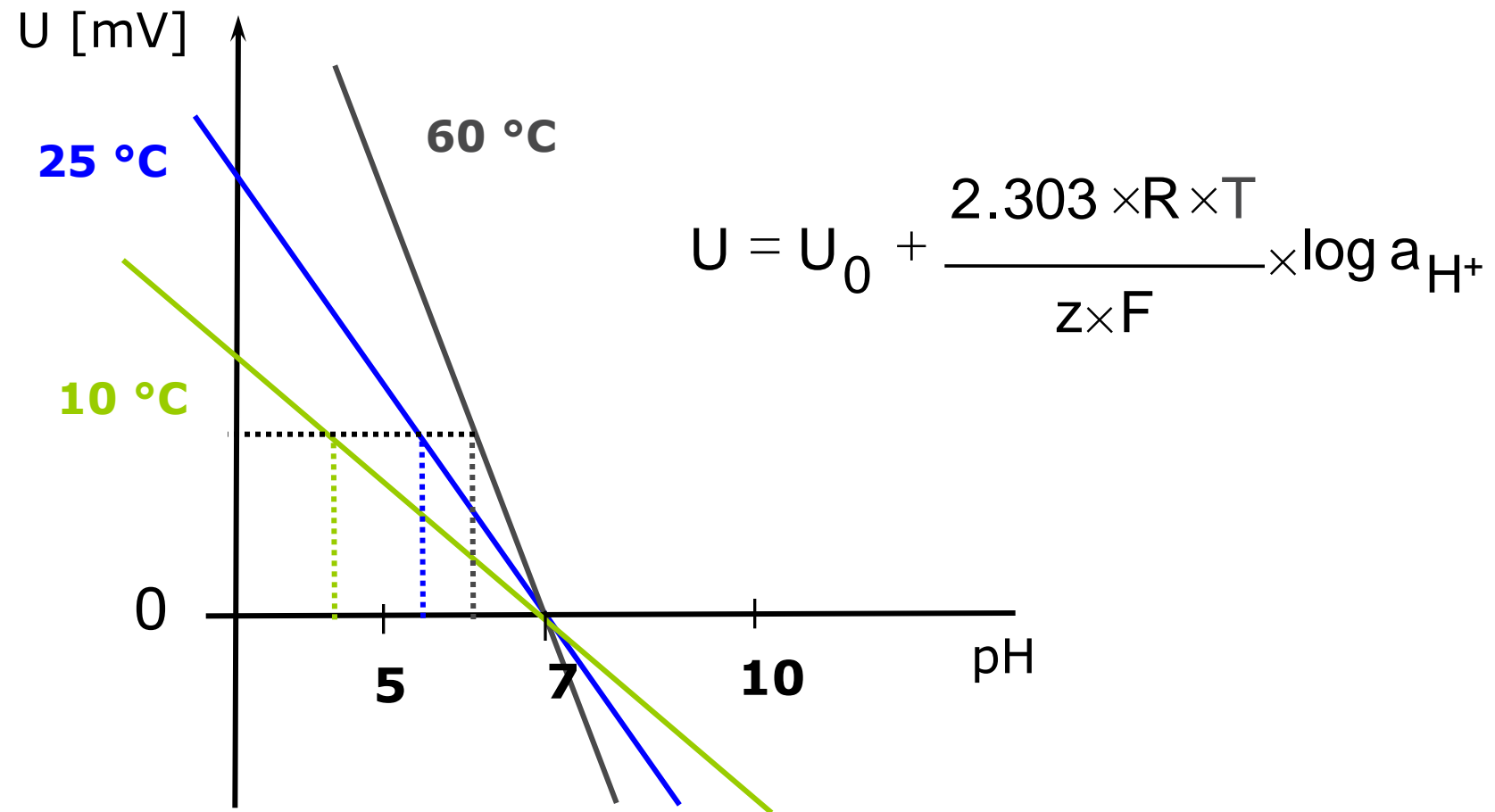
°C	pH
0	7.11
5	7.08
15	7.04
20	7.02
25	7.00
30	6.99
38	6.98
40	6.98
50	6.97
60	6.97
70	6.98
80	7.00
90	7.01

T [°C]	pH
	±0.02
0	7.12
5	7.09
10	7.06
15	7.04
20	7.02
<b>25</b>	<b>7.00</b>
30	6.99
35	6.98
40	6.97
45	6.97
50	6.97

Temp. °C	pH	Temp. °C	pH
0	7,13	35	6,96
5	7,07	40	6,85
10	7,05	50	6,95
15	7,02	60	6,96
20	7,00	70	6,96
25	6,98	80	6,97
30	6,98	90	7,00



# Calibration: Influence of the temperature



# Temperature compensation

pH meter can only correct the temperature behaviour of the electrode and never of the sample

e.g. 0.001 M NaOH

0 °C	25 °C	50 °C
11.94	11.00	10.26

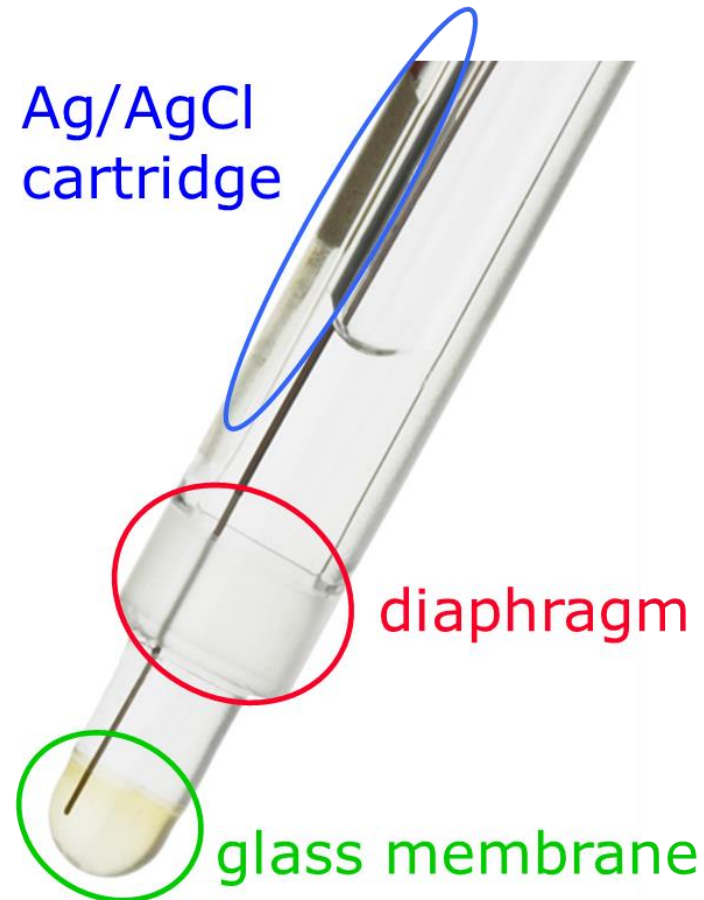
Measurement uncertainty increases with temperature difference calibration to measurement

→ Recommendation:

$$T_{\text{meas}} = T_{\text{cal}}$$

→ **always measure your samples at the same temperature if you want to compare their pH values.**

# pH Regulation Liquefaction & Fermentation



## Electrode Maintenance

RINSE – DI WATER, ETHANOL, IPA

DO NOT WIPE OR RUB GLASS MEMBRANES

REFERENCE JUNCTION (DIAPHRAGM)

REFERENCE FILLED (KCL)

CALIBRATION

VIDEOS:

[HTTPS://WWW.METROHM.COM/EN\\_US/SERVICE/PRODUCT-HELP-CENTER/ELECTRODES/MAINTENANCE-AND-CARE-OF-ELECTRODES.HTML](https://www.metrohm.com/en_us/service/product-help-center/electrodes/maintenance-and-care-of-electrodes.html)

[HTTPS://WWW.METROHM.COM/EN/DISCOVER/BLOG/2024/CALIBRATE-PH-METER.HTML#SLOPE-OFFSET](https://www.metrohm.com/en/discover/blog/2024/calibrate-ph-meter.html#slope-offset)

# Thank You!

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